SYNTHESIS OF THREE NO-CARRIER-ADDED 06-4-[1"1] IODOBENZYLGUANOSINE DERIVATIVES, NEW REAGENTS FOR

THE ASSAY OF O~ALKYLGUANINE-DNA ALKYLTRANSFERASE ACTIVITY

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SUMMARY

06-alkylguanine-DNA alkyltransferase (AGT) is mainly responsible for tumour resistances observed in chemotherapeutic treatments by chloroethylnitrosoureas (CENUs). Measurement of AGT activity **is** thereby essential to predict the response of the patients to therapy with CENUs. In order to develop a sensitive and easy new assay for AGT, previously undescribed **06-4-[1251]iodobenzyl-2'-deoxyguanosine, 06-4-** [**'zI]iodobenzyl-N-acetylguanosine** and **O6-4-[IzI]** iodobenzylguanosine labelled with **high** specific activity were prepared. The most convenient synthetic **route** appeared to be a rapid and high yield iododestannylation of a tri-n-butylstannyl derivative with no-carrieradded sodium $\left[\begin{smallmatrix} 125\\1\end{smallmatrix}\right]$ iodide. Final HPLC separation from the excess of precursor and unreacted [¹²⁵I] iodides afforded the radioiodinated guanosine derivatives in yields ranging from **70** to **77** %, chemical and radiochemical purities averaging 99%.

Key words : O^6 -4-[¹²⁵I]iodobenzyl-2'-deoxyguanosine, O^6 -4-[¹²⁵I]iodobenzyl-N-acetylguanosine, O⁶-4-^{[125}I] iodobenzylguanosine, iododestannylation, assay for AGT activity.

INTRODUCTION

Akylating agents, such **as** chloroethylnitrosoureas (CENUs), used for cancer chemotherapy

are known to generate O⁶-alkylguanine adducts leading to DNA damage mainly responsible for their

CCC 03624803/95/121215-1 ^I 01995 **by John Wiley** & **Sons, Ltd.** **Received 7 July 1995 Revised 12 July 1995** cytotoxic activity **(1-3).** Such lesions can be repaired by the action of a protein called *06* alkylguanine-DNA alkyltransferase (AGT) which transfers the 06-alkyl group to **an** internal cysteine residue. By this irreversible reaction, the AGT is inactivated and native guanine restored in DNA **(4-7).**

Therefore, the capacity of cells to repair O^6 -alkylguanine adducts, resulting in turnour cell resistance to alkylating agents, is directly correlated to AGT content of cells (8). **As** the constitutive level of AGT varies considerably with the species and cell type. measurement of AGT activity is of fundamental importance for clinical applications to predict whether a chemotherapeutic treatment by chloroethylating agents will be effective.

Many different methods have already been developed for assaying AGT activity. The commonest procedure measures the difference between the amount of initial and residual O⁶-¹³HI methylguanine DNA substrate when incubated with AGT (9). Another similar approach uses $\left[^{32}P\right]$ or **f35S]** labelled oligodeoxynucleotide containing 06-methylguanine **as** a substrate for AGT **(10,ll).** Nevertheless, these methods raise substantial problems : time consuming preparation of the substrate and separation after incubation unsuitable for large predictive tests, rapid degradation of the substrate and short half-life of the radionuclide **requiring** frequent resynthesis, semiquantitative **assays.**

For the purpose of developing a rapid, **easy** and sensitive method for direct measurement of AGT activity, we have considered a new approach. The theoretical basis of our assay lies in **the** AGT mechanism of action. Since AGT repairs O⁶-alkylguanine lesions, it could irreversibly transfer an O⁶- $4\cdot \int_0^{125}$ II iodobenzyl group from an O^6 -4- I^{125} II iodobenzylguanosine derivative, substrate for AGT, to its cystein residue resulting in its radiolabelling, making its quantification possible. Also, the use of **06-4-['251]** iodobenzylguanosine derivatives, **as** free nucleosides or incorporated in DNA, with high specific activity should lead to a very sensitive and direct detection of the protein.

In this paper we report a rapid and straight forward preparation of the $O⁶-4-[125]$ [iodobenzyl-2'deoxyguanosine **(8),** Q^6 -4- I^{125} Thiodobenzyl-N-acetylguanosine **(9)** and Q^6 -4- I^{125} Tl iodobenzylguanosine (10), all of them substrates for AGT, labelled with high specific activity from their **06-4-(tri-n-butylstannyl)benzyl** precursors.

SYNTHESIS

Scheme 1 : Synthesis of O^6 -4- I^{125} ^T iodobenzylguanosine derivatives by iododestannylation.

The reaction pathway followed to prepare the O^6 -4- $[$ ¹²⁵I] iodobenzylguanosine derivatives **8**. **2,** is outlined in Scheme 1. **06-4-bromobenzylperacetyl~2'-deoxyguanosine 1** and guanosine **2,** as starting materials, were synthesized by O^6 -4-bromobenzylation of peracetyldeoxyguanosine and guanosine respectively according to a method previously described (12). The reaction of these aromatic bromides with excess hexabutylditin in the presence of **tetrakis(tripheny1phosphine)palladium as** catalyst, in refluxing toluene, afforded compounds **9** and **4** in 51% and 58% yields (13). The tri-n-butylstannyl intermediates 3 and 4 were completely or partially de-acetylated in pyridine under alkaline conditions (NaOH) within hours or minutes respectively to obtain stable **06-4-(tri-n-butylstannyl)benzyl-2'-deoxyguanosine** *0.* 06-4-(tri-nbutylstannyl)benzylguanosine (7) and O⁶-4-(tri-n-butylstannyl)benzyl-N-acetylguanosine (6) in good yields. These precursors undergo ready iododestannylation upon treatment with no-carrier-added sodium \int_1^{125} I] iodide in the presence of chloramine T in aqueous ethanol, for 30 minutes at room temperature, yielding the desired O^6 -4- $[$ ¹²⁵I] iodobenzylguanosine derivatives **8**, **2**, **10** (14). Iodinated product isolation from the tri-n-butylstannyl precursor and unreacted $\int^{125} \Pi$ iodides, followed by determination of chemical and radiochemical purities were performed respectively by semi-preparative and analytical reverse phase high performance liquid chromatography (HPLC) monitored by both UV and gamma detection using a gradient of methanol/water as mobile phase (Figure 1). Isolated radiochemical yields ranged from 70% to 77%. chemical **and** radiochemical purities averaging **99%.**

Time (min)

: **HPLC** elution profile of a crude $O^6-4-[1^{125}I]$ iodobenzylguanosine derivative. Chromatogram **(1)** UV detection : Rt = **38.46** tri-n-butylstannyl precursor ; *(2)* radioactivity detection : $Rt = 3.20$ unreacted $\begin{bmatrix} 1^{25} \end{bmatrix}$ iodides, $Rt = 23.92$ $\begin{bmatrix} 1^{25} \end{bmatrix}$ iodinated product. Operating conditions (system **A) are** reported in Experimental Section.

All the reactions were initially canied out **using** non-radioactive sodium iodide. The final compounds were characterized by proton nuclear magnetic resonance, mass spectrometry and infrared spectroscopy. Each non-radioactive compound **was** used **as** an authentic standard in the HPLC analysis to identify and confirm the product of interest.

In addition, a radioanalytical quality control has **was** performed by HPLC over 3 months. Each iodinated guanosine derivative proved to be stable. No significant deiodination was detected.

DISCUSSION

The results reported here suggest that iododestannylation is a suitable method for the preparation of **06-4-['2511** iodobenzylguanosine derivatives labelled with high specific activity. The major advantages of this approach **are** that the labelling procedure may be effected quickly, in the last step of the synthetic route, with good yields and under mild conditions. Furthermore, the final radiolabelled product can be cleanly separated from the starting materials and its specific activity depends only on the activity of the iodinating species, allowing high specific activity labelling.

EXPERIMENTAL

General comments. Sodium [¹²⁵I] iodide as no-carrier-added solution in reductant-free aqueous sodium hydroxide was purchased from CIS bio international. *AU* chemicals were from commercial suppliers **and** used **as** received. Proton nuclear magnetic resonance (H-NMR) **spectra** were **1** performed on Briicker AM 200 (4.5") spectrometer. Chemical shifts **(8) are** reported in parts per million relative to the internal tetramethylsilane standard. Electron impact mode **@I)** mass **spectra** (MS) were obtained on a Hewlett Packard **5989** A instrument Infrared (IR) **spectra** were recorded on a Perkin Elmer 398 spectrometer. Melting points (mp) were determined on an Electrothermal digital apparatus. Analytical thin layer chromatography (TLC) was conducted on precoated silca gel plates (Merck 60F₂₅₄, 0.2 mm thick and Merck RP 18 F₂₅₄S, 0.25 mm thick) with both detection by ultra violet light **at** 254 nm and visualization by iodine. Silica gel 60 (Chromagel, **230-400** mesh, SDS) was used for medium pressure chromatography using the indicated solvent mixture expressed as volumdvolume ratios. HPLC pwifcation was performed on a Shimadzu HPLC system (LC6A pump, SCL6B system controller, CR5A integrator) equipped with a semi-preparative reverse phase column (Lichroprep RP **18,** 25-40 pm. 200x12 mm), connected to a Shimadzu SPD6AV *UV* spectrophotometric detector (254 nm) in series with a Raytest NaI (Tl) gamma detector (system A). The HPLC analytical system consisted of a Chromatem **800,** Pye Unicam PU 4020 (Philips) *UV* detector and **a** Radiomatic *A200* radioactivity flow one detector employing a reverse phase column (Spherisorb **WC 18, 5** ym. 250~4.6 mm) (system B). The solvent mixture indicated later **are** expressed as volume/volume ratios. Radioactive samples were measured using a Packard 5530 auto gamma counter.

To a solution of O^6 -4-bromobenzylperacetyl-2'-deoxyguanosine 1 (1.8 g, 3.2 mmol) or guanosine **2** (2.0 g, 3.2 mmol) in toluene (32 mL) was added hexabutylditin (3.7 g, 16.1 mmol) and a catalytic amount of **tetrakis(tripheny1phosphine)palIadium** (42 mg). The mixture was stirred at reflux **temperature** for 24 h under nitrogen atmosphere, filtered. and the **filtrate** evaporated under **reduced** pressure. The crude product was purified by chromatography on silica gel eluting with ethyl acetate/hexane (70/30).

06-4-(tri-n-butylstannyl)benzylperacetyl-2~-deo~yguanosine (3) .

Yield : 51%;

TLC (ethyl acetate/hexane, $80/20$) $Rf = 0.50$;

IR (KBr) $v : 2850-3000 \text{ cm}^{-1}$ (CH, CH₂, CH₃), 1755 cm⁻¹ (C=O, ester), 1700 cm⁻¹ (C=O, amide), 1230 cm^{-1} (=C-O):

¹H-NMR (CDCl₃) δ : 0.87 (t, 3H, CH₃), 1.00-1.08 (m, 2H, SnCH₂CH₂CH₂CH₃), 1.22-1.39 (m, 2H, SnCH₂CH₂CH₂CH₃), 1.45-1.60 (m, 2H, SnCH₂CH₂CH₂CH₃), 2.08, 2.15 (2s, 6H, 2 COCH₃), 2.53 (s, 3H, NCOCH₃), 2.55-2.62 and 2.94-3.05 (2m, 2H, H₂,), 4.33-4.44 (m, 3H, H₄,, H₅.), 5.42-5.45 (m, 1H, H₃.), 5.58 (s, 2H, $CH_2C_6H_4$), 6.35 (d, 1H, H₁.), 7.44 (s, 4H, C₆H₄), 7.97 (s, 1H, H_g), 7.99 (s, 1H, NH, exchanges with D_2O).

06-4-(tri-n-butylstannyl)benzylperacetylguanosine (4).

 $Yield : 58\%; mp 70-72°C$;

TLC (ethyl acetate/hexane, $80/20$) $Rf = 0.43$;

IR (KBr) $v : 2850-3000 \text{ cm}^{-1}$ (CH, CH₂, CH₃), 1740 cm⁻¹ (C=O, ester), 1670 cm⁻¹ (C=O, amide), 1230 cm^{-1} (=C-O);

¹H-NMR (CDCl₃) δ : 0.87 (t, 3H, CH₃), 1.04 (t, 2H, SnCH₂CH₂CH₂CH₃), 1.26-1.37 (m, 2H, $SnCH_2CH_2CH_2CH_3$), 1.45-1.61 (m, 2H, $SnCH_2CH_2CH_2CH_3$), 2.09 and 2.14 (2s, 6H, 2

COCH₃), 2.51 (s. 3H, NCOCH₃), 4.42-4.45 (m, 3H, H₄₁, H₅), 5.58 (s, 2H, $CH_2C_6H_4$), 5.87-6.00 (m, 1H, H₃), 5.90-5.92 (m, H, H₂), 6.03 (d, 1H, H₁), 7.45 (s, 4H, C₆H₄), 7.90 (s, 1H, H8), 7.94 **(s.** 1H. **NH,** exchanges with **D,O).**

General procedure for the preparation of O^6 -4-(tri-n-butylstannyl)benzyl-2'**deoxyguanosine** &) **and O6-4-(tri-n-butylstanny1)benzyIguanosine** *0.*

A solution of the relevant compound **3** (1.5 **g,** 2 mmol), **4** (1.7 g, 2 mmol) in pyridine (9 mL) was treated with NaOH 2.5 M (3.6 mL, 9 mmol for 2'-deoxyguanosine derivative **and** 4.8 mL, 12.0 mmol for guanosine derivative) and stirred for 2 h at 30'C. Acidic cation exchanger (Dowex 50x8 resin, 12 mL) was added with vigorous stirring, filtered, washed with **a minimum** amount of pyridine. The solution was evaporated under reduced pressure. The residue was then crystallized in hexane to give a white solid.

06-4-(tri-n- butylstannyl)benzyl-2'-deoxyguanosine *0* .

Yield : 95%; mp 147-149°C;

TLC (dichoromethane/ethanol, $90/10$) $Rf = 0.38$, TLC (ethanol/water, $80/20$) $Rf = 0.17$;

IR (KBr) v : 2850-3000 cm⁻¹ (CH, CH₂, CH₃), 1230 cm⁻¹ (=C-O);

¹H-NMR (CDCl₃) δ : 0.85 (t, 3H, CH₃), 1.00–1.05 (m, 2H, SnCH₂CH₂CH₂CH₃), 1.23-1.34 (m, 2H, SnCH₂CH₂CH₂CH₃), 1.43-1.54 (m, 2H, SnCH₂CH₂CH₂CH₃), 2.24-2.30 (m, 2H, OH₃, H_2), 2.90-3.05 (m, 1H, H_2), 3.75, 4.00 (dd, 2H, H_5), 4.19 (m, 1H, H_4), 4.75 (d, 1H, H_4), 5.18 (br s, 2H, NH₂, exchange with D₂O), 5.51 (s, 2H, $CH_2C_6H_4$), 6.18-6.23 (m, 1H, H₁), 6.75 (m, 1H, OH₅), 7.41 (s, 4H, C₆H₄), 7.63 (s, 1H, H₂).

06-4-(tri-n-butylstannyl)benzylguanosine (IL) .

Yield: 98%; mp 75-77°C;

TLC (dichoromethane/ethanol, $90/10$) $Rf = 0.38$, TLC (ethanol/water, $80/20$) $Rf = 0.30$;

IR (KBr) \vee : 2850-3000 cm⁻¹ (CH, CH₂, CH₃), 1230 cm⁻¹ (=C-O);

¹H-NMR (DMSO-d₆) δ : 0.85 (t, 3H, CH₃), 1.03 (t, 2H, SnCH₂CH₂CH₂CH₂CH₃), 1.23-1.38 (m, 2H, SnCH₂CH₂CH₂CH₃), 1.46-1.55 (m, 2H, SnCH₂CH₂CH₂CH₃), 3.57-3.62 (m, 2H, H₅¹), 3.88-3.90 (m, 1H. **H4),** 4.11 **(t,** lH, H,,), 4.45 **(t,** H. H,,), 5.12 (m. 2H. OH,. **OHyh** 5.47 **(S.** 3H. OH₂, CH₂C₆H₄), 5.80 (d, 1H, H₁), 7.45 (s, 4H, C₆H₄), 6.49 (s, 2H, NH₂, exchange with D₂O), 8.12 (s, $1H, H_o$).

Synthesis of 06-4-(tri-n-butylstannyl)benzyl-N-acetylguanosine (6) .

A solution of **4** (1.7 g. 2 mmol) in pyridine (9 mL) was treated with **NaOH** 2.5 M (2.4 mL, *6* mmol) and stirred for 6 min. The mixture was neutralized by adding acidic cation exchanger (Dowex 50x8 resin, 8 mL) with vigorous stirring, filtered, washed with a minimum amount of pyridine. "he solution was evaporated under reduced pressure. The residue was chromatographed on silica gel with a gradient of dichloromethane and dichloromethane/ethanol (98/2, 95/5, 90/10) as eluent to give a white solid.

 $Yield : 83\%; mp 75-77°C$;

TLC (dichoromethane/ethanol, $90/10$) $Rf = 0.54$, TLC (ethanol/water, $80/20$) $Rf = 0.39$;

IR (KBr) $v : 2850-3000 \text{ cm}^{-1}$ (CH, CH₂, CH₃), 1230 cm⁻¹ (=C-O);

¹H-NMR (DMSO-d₆) δ : 0.92 (t, 3H, CH₃), 1.12 (t, 2H, SnCH₂CH₂CH₂CH₃), 1.25-1.38 (m, 2H, SnCH₂CH₂CH₂CH₃), 1.52-1.63 (m, 2H, SnCH₂CH₂CH₂CH₃), 2.25 (s, 3H, NCOCH₃), 3.51-3.63 (m, 2H, H₅⁾, 3.88-3.93 (m, 1H, H₄⁾, 4.18 (t, 1H, H₃⁾, 4.57 (t, H, H₂⁾, 4.96 (t, 1H, OH₅⁾, 5.16 (d, 1H, OH₃), 5.45 (d, 1H, OH₂), 5.57 (s, 2H, $CH_2C_6H_4$), 5.88 (d, 1H, H₁), 7.42-7.52 (m, 4H, C_6H_4), 8.44 (s, 1H, H_a), 10.47 (s, 2H, NH, exchanges with D_2O).

General procedure for the preparation of O^6 -4-iodobenzyl-2'-deoxyguanosine (nonradioactive $\hat{\mathbf{g}}$), $\mathbf{0}^6$ -4-iodobenzyl-N-acetylguanosine (non-radioactive $\hat{\mathbf{g}}$) and $\mathbf{0}^6$ -4iodobenzylguanosine (non-radioactive 10).

To a solution of the relevant tri-n-butylstannyl precursor $\frac{5}{2}$, $\frac{6}{2}$ or $\frac{7}{2}$ (0.6 mmol, 0.4 g) in ethanol was added sodium iodide $(1.3 \text{ mm}$ mol, $0.2 \text{ g})$ and chloramine T $(1.3 \text{ mm}$ ol, $0.2 \text{ g})$ in aqueous solution (15mL). After 30 **min** at room temperature, the excess oxidizing agent was reduced by the addition of *5%* aqueous sodium metabisulfite (20.3 mL).

06-4-iodobenzyl-2'-deoxyguanosine (non-radioactive *s).*

Yield: 60%; mp 127-129°C;

TLC(dichloromethane/ethanol, $90/10$) $Rf = 0.52$, TLC (ethanol/water, $80/20$) $Rf = 0.75$;

IR (KBr) v : 3440, 3320 cm⁻¹ (NH₂), 1230 cm⁻¹ (=C-O);

¹H-NMR (DMSO- d_6) δ : 2.20, 2.53 (m, 2H, H₂), 3.50-3.55 (m, 2H, H₅), 3.80 (m, 1H, H₄), 4.34 $(m, 1H, H_1)$, 4.99 (t, 1H, OH₅^t, exchanges with D₂O), 5.27 (d, 1H, OH₃^t, exchanges with D₂O), 5.43 **(s, 2H,** $CH_2C_6H_4I$ **), 6.19 (t, 1H, H₁**), 6.49 **(s, 2H, NH₂, exchange with D₂O), 7.27-7.76** (dd, 4H, C_6H_4I), 8.08 (s, 1H, H_o);

 $MS m/z$:: 483 (M), 367 (B + H), 241 (B + 2H -I), 240 (B + H - I), 217 (CH₂C₆H_AI), B = O^6 -4iodobenzylguanine - H).

06-4-iodobenzyl.N-acetylguanosine (non-radioactive e>.

 $Yield : 75\% ; mp 146-148°C ;$

TLC(dichloromethane/ethanol, 90/10) $Rf = 0.28$, TLC (ethanol/water, 80/20) $Rf = 0.75$;

IR (KBr) v : 3500-3200 cm⁻¹ (OH), 1670 cm⁻¹ (C=O amide), 1230 cm⁻¹ (=C-O);

¹H-NMR (DMSO- d_6) δ : 2.24 (s, 3H, CH₃CO), 3.53-3.64 (m, 2H, H₅¹), 3.90-3.94 (m, 1H, H₄¹),

4.13-4.19 (m, 1H, H₃), 4.50-4.59 (m, 1H, H₂), 4.96 (t, 1H, OH_{5'}, exchanges with D₂O), 5.17 (d,

1H, OH_{3} , exchanges with D_2O), 5.46 (d, 1H, OH_{2} , exchanges with D_2O), 5.56 (s, 2H, CH~CL@~I), 5.88 (d, 1H, **HI,),** 7.33-7.76 (dd, 4H, C6H41), **8.44 (s, 1H,** H8), 10.47 **(s,** lH, NH, exchanges with D_2O ;

MS *m/t* : 541 (M), 452 (B + **44),** 409 (B + H), 367 (B + 2H - COCH,), 240 (B + 2H - COCH,-I), 217 **(CH₂C₆H₄I). B =** O^6 **-4-iodobenzyl-** N^2 **-acetylguanine - H).**

O⁶-4-iodobenzylguanosine (non-radioactive 10).

Yield : 80% ; mp 207-209°C;

TLC(dichloromethane/ethanol, $90/10$) $Rf = 0.30$, TLC (ethanol/water, $80/20$) $Rf = 0.82$; IR (KBr) v : 3440, 3320 cm⁻¹ (NH₂), 1230 cm⁻¹ (=C-O);

¹H-NMR (DMSO- d_6) δ : 3.56-3.61 (m, 2H, H₃), 3.90 (m, 1H, H₄), 4.10 (m, 1H, H₃), 4.44 (m, 1H, H₂,), 5.13 (m, 2H, OH₃,, OH₅[,], exchange with D₂O), 5.43 (m, 3H, OH₂[,], exchanges with D₂O, $CH_2C_6H_4I$, 5.78 (d, 1H, H₁,), 6.50 (s, 2H, NH₂, exchange with D₂O), 7.28-7.77 (dd, 4H, C_6H_4I , 8.11 (s, 1H, H_6);

 \overline{MS} *m/z* : : 499 (M), 367 (B + H), 241 (B + 2H - I), 240 (B + H - I), 217 (CH₂C₆H₄I), B = O^6 -4iodobenzylguanine - H).

General procedure for the preparation of **06-4-['251]iodobenzyI-2'.deoxyguanosine (a, 06-4-[1251]iodobenzyI-N-acetylguanosine** *Ce,* and **06-4-['251]** iodobenzylguanosine $(10).$

To a 10 mg/mL solution of the relevant tri-n-butylstannyl precursor 5, 6 or 7 (3 µmol, 200 pL) in ethanol placed in a conical reaction vial (Wheaton, 1mL) fitted with a septum was added sodium $\int_0^{125} \text{I}$ iodide (50 µL, 500 µCi, 18.5 MBq) followed by a 10 mg/mL chloramine T aqueous solution $(2 \mu \text{mol}, 50 \mu \text{L})$. After 30 min at room temperature, the excess oxidizing agent was reduced by the addition of a **15** mg/mL sodium metabisulfite aqueous solution **(4** pmol, 50 pL). The material was purified by reverse phase HPLC at a *5* mIJmin flow **rate** using as mobile phase a gradient of ethanol/water mixture beginning at 36/64, increasing to 40/60 from 5 min to 25 min, and then to **80120** over **55** min (system A).

Determination of chemical and radiochemical final purities was carried out by analytical HPLC at a 1.5 mL/min flow rate using as mobile phase a gradient methanol/water mixture at $40/60$ (system B). Each radioiodinated product 8, 9, 10 co-migrated with non radioactive standard and showed chemical and radiochemical purities averaging 99%.

$0⁶-4-1¹²⁵$ Iliodobenzyl-2'-deoxyguanosine (8) .

Yield based on starting sodium $\int_0^{125} \Pi$ iodide : 78% (390 µCi, 14.5 MBq);

Rt (system A) = 23.9 min, Rt (system B) = 8.6 min.

06-4-['251]iodobenzyI-N-acetylguanosine @).

Yield based on starting sodium \int_0^{125} **[**] iodide : 70% (350 µCi, 13.0 MBq);

Rt (system A) = 20.0 min, Rt (system B) = 5.6 min.

O^6 -4- $\binom{125}{1}$ **iodobenzylguanosine** (10).

Yield based on starting sodium \int_0^{125} I] iodide : 73% (365µCi, 13.5 MBq);

Rt (system A) = 22.4 min, Rt (system B) = 6.6 min.

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